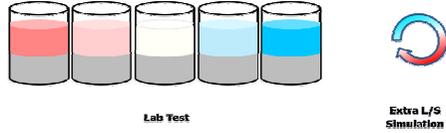


● Industrial site composite
 ○ own pH Industrial site composite
 --- Model Material, L/S=11.1(P,1,1)
 ○ own pH model at L/S=11
 --- Model Material, L/S=0.40(P,1,1)
 ○ own pH model at L/S=0.4
 --- DTL

Object
Name

pH Dependent Leaching Test Model
Industrial F contaminated site

pH Dependent Leaching Test Scenario



Lab Test

Model Parameters

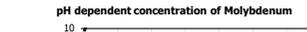
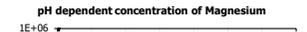
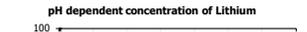
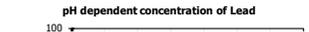
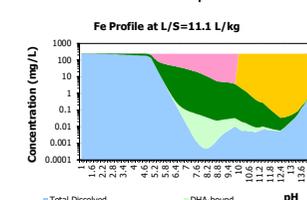
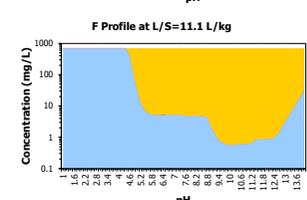
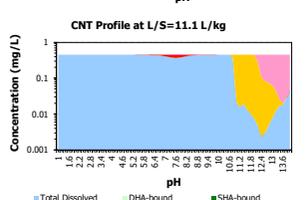
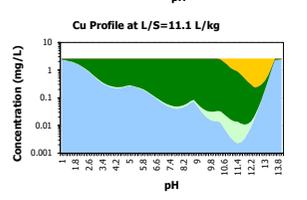
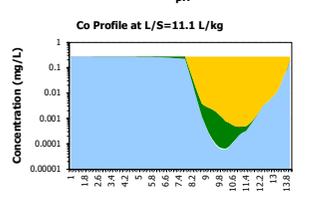
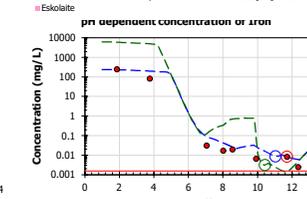
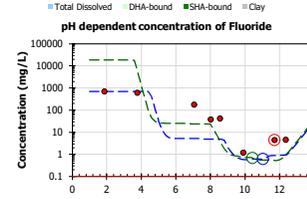
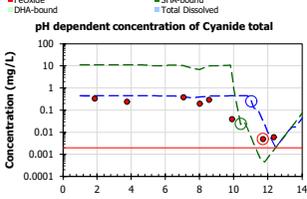
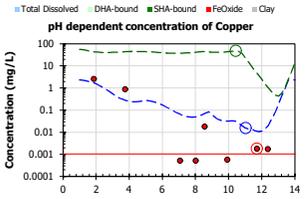
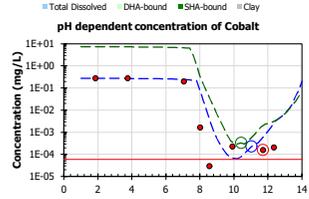
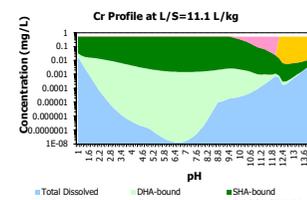
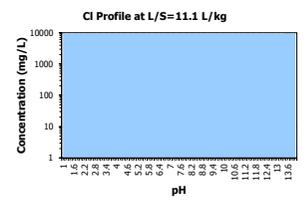
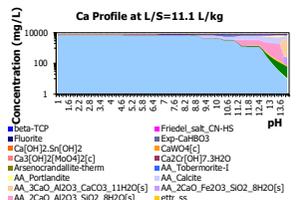
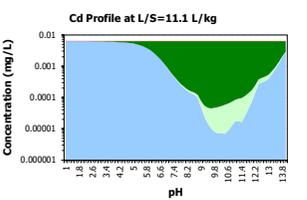
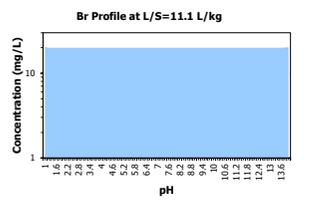
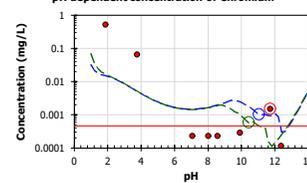
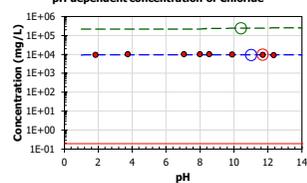
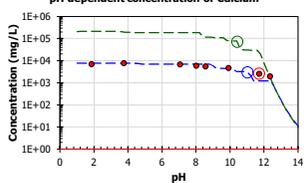
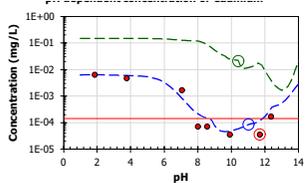
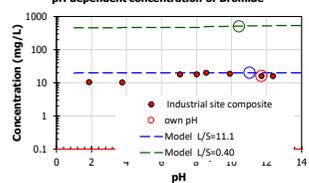
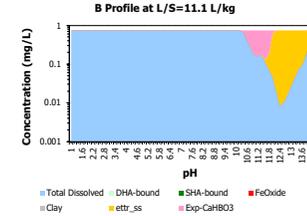
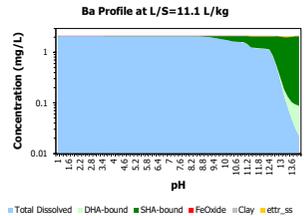
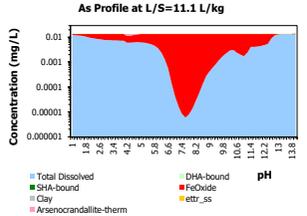
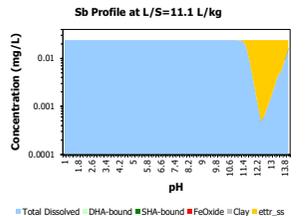
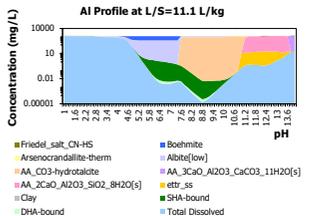
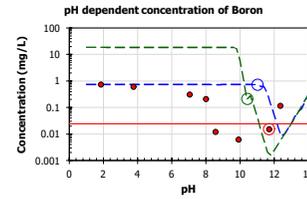
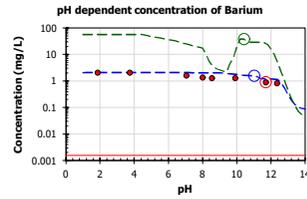
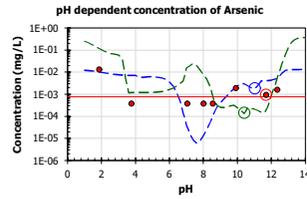
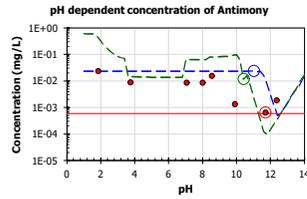
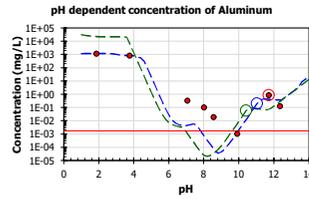
Entity	Unit	Default	Available Content	Entity	mg/kg	Entity	mg/kg	Entity	mg/kg
c0		-5.030	Al	1.319E+04	Fe	2728	PO4	40.75	
c1		0.7637	As	0.1499	B	8.271	Sb	0.2662	
c2		-0.3660	Ba	23.40	Si	1.271E+04	Se	0.02844	
c3		0.05515	Br	221.7	Hg	2.006E-07	Sn	1.069	
c4		-0.003379	Ca	8.557E+04	K	1.524E+04	SO4	929.5	
c5		7.446E-05	Cd	0.07052	Li	23.40	Sr	53.48	
Clay	mg/kg	1.900E+04	Cl	1.046E+05	Mg	1.860E+05	Th	2.320E-07	
Hydrous Ferric Oxide	mg/kg	180.0	CNT	5.001	Mn	390.3	U	2.380E-07	
L/S	L/kg	11.11	Co	3.075	Mo	0.6076	V	0.3282	
pE		-0.5000	CO32-	5.010E+04	Na	2.221E+04	W	1.0000	
pH		11.50	Cr	5.834	Ni	13.31	Zn	75.39	
Solid Humic Acid	mg/kg	2.160E+04	Cu	28.61	NO3	210.0			
Simulated Low L/S	L/kg	0.4000	F	7784	Pb	9.780			

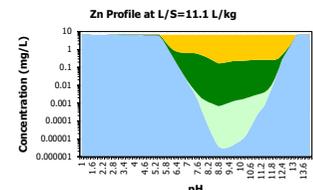
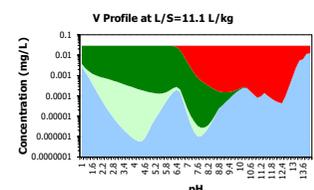
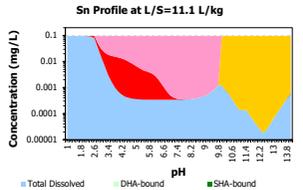
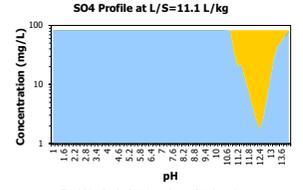
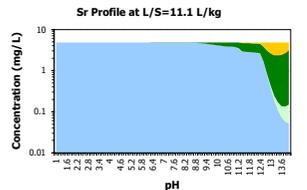
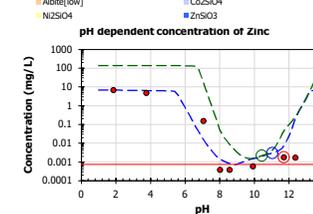
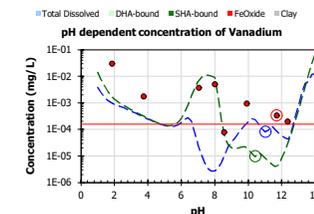
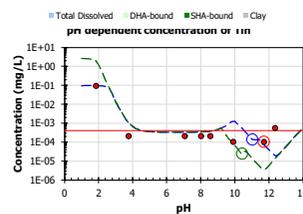
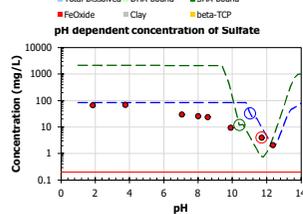
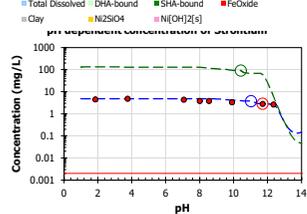
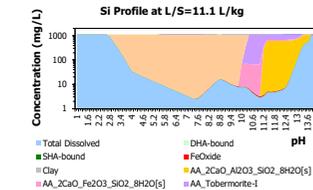
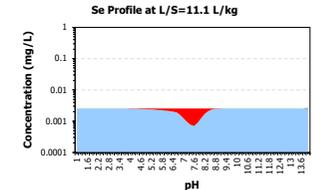
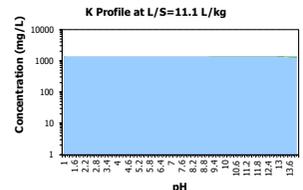
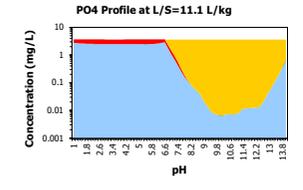
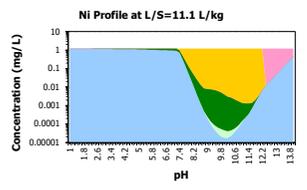
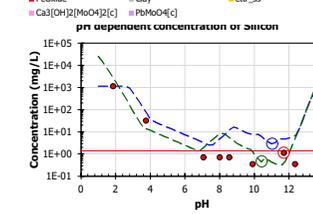
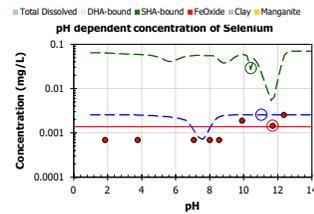
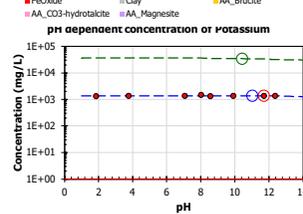
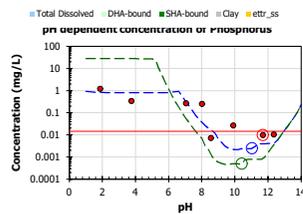
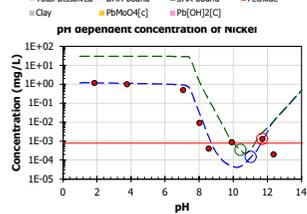
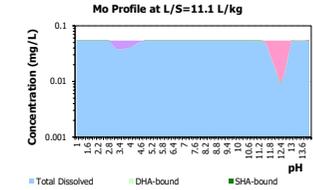
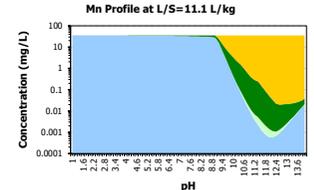
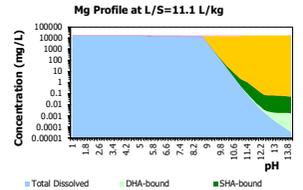
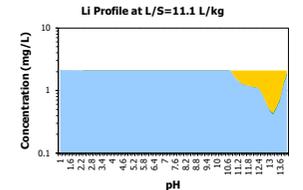
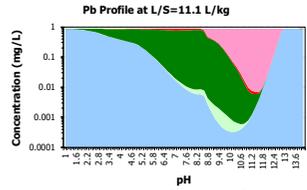
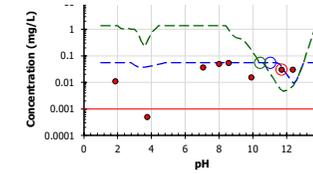
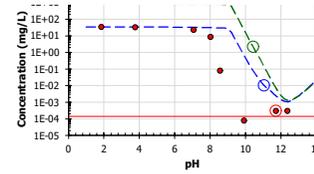
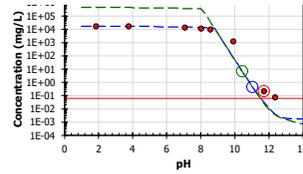
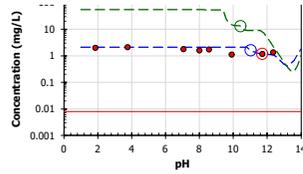
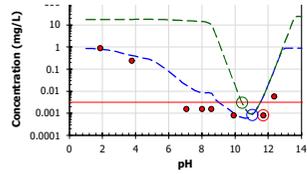
Solid Solutions

Name	End Member	Log(K) Reaction
ettr_ss	AsO4_Ettringite_ss	26.79 AsO4_Ettringite_ss + 1 H+ + 8 H2O -> 2 Al[OH]4- + 3 AsO4-3 + 6 Ca+2 + 1 ettr_ss
	Ba_Ettringite_ss	4.008 Ba_Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ba+2 + 3 SO4-2 + 1 ettr_ss
	BO3_Ettringite_ss	-46.87 BO3_Ettringite_ss + 7 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 H2BO3- + 1 ettr_ss
	CrO4_Ettringite_ss	-8.592 CrO4_Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 CrO4-2 + 1 ettr_ss
	Ettringite_ss	-10.99 Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 SO4-2 + 1 ettr_ss
	Li-Ettringite_ss	-5.699 Li-Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 5 Ca+2 + 2 Li+ + 3 SO4-2 + 1 ettr_ss
	MoO4_Ettringite_ss	-9.592 MoO4_Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 MoO4-2 + 1 ettr_ss
	Sb[OH]6_Ettringite_ss	-33.80 Sb[OH]6_Ettringite_ss + 7 H+ + 17 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 Sb[OH]6- + 1 ettr_ss
	SeO4-2_Ettringite_ss	4.408 SeO4-2_Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 SeO4-2 + 1 ettr_ss
	Sn[OH]4-2_Ettringite_ss	-69.18 Sn[OH]4-2_Ettringite_ss + 8 H+ + 4 H2O -> 2 Al[OH]4- + 6 Ca+2 + 2 SO4-2 + 1 Sn+2 + 1 ettr_ss
	Sr_Ettringite_ss	4.008 Sr_Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 3 SO4-2 + 6 Sr+2 + 1 ettr_ss
	VO3_Ettringite_ss	-53.79 VO3_Ettringite_ss + 13 H+ + 2 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 VO2+ + 1 ettr_ss
	WO4_Ettringite_ss	-9.592 WO4_Ettringite_ss + 4 H+ + 8 H2O -> 2 Al[OH]4- + 6 Ca+2 + 3 WO4-2 + 1 ettr_ss

Minerals

Name	Log(K)	Reaction	Name	Log(K)	Reaction
AA_2CaO_Al2O3_SiO2_8H2O[s]	18.18	AA_2CaO_Al2O3_SiO2_8H2O[s] -> 2 Al[OH]4- + 2 Ca+2 + 3 H2O + 1 H2SiO4-2	CaMoO4[c]	7.940	CaMoO4[c] -> 1 Ca+2 + 1 MoO4-2
AA_2CaO_Fe2O3_SiO2_8H2O[s]	22.49	AA_2CaO_Fe2O3_SiO2_8H2O[s] -> 2 Ca+2 + 2 Fe[OH]4+ + 3 H2O + 1 H2SiO4-2	CaSb[OH]6[s]2	18.41	CaSb[OH]6[s]2 + 6 H2O -> 1 Ca+2 + 2 Sb[OH]6-
AA_3CaO_Al2O3_CaCO3_11H2O[s]	-24.52	AA_3CaO_Al2O3_CaCO3_11H2O[s] + 4 H+ -> 2 Al[OH]4- + 1 CO3-2 + 4 Ca+2 + 9 H2O	CaWO4[c]	9.000	CaWO4[c] -> 1 Ca+2 + 1 WO4-2
AA_Brucite	-16.84	AA_Brucite + 2 H+ -> 2 H2O + 1 Mg+2	Co2SiO4	6.289	Co2SiO4 + 2 H+ -> 2 Co+2 + 1 H2SiO4-2
AA_Calcite	9.481	AA_Calcite -> 1 CO3-2 + 1 Ca+2	Cuprite	6.990	Cuprite + 2 H+ -> 2 Cu+2 + 1 H2O + 2 e-
AA_CO3-hydroxalcite	-4.852	AA_CO3-hydroxalcite + 4 H+ -> 2 Al[OH]4- + 1 CO3-2 + 6 H2O + 4 Mg+2	Eskolaite	139.5	Eskolaite + 5 H2O -> 2 CrO4-2 + 10 H+ + 6 e-
AA_Magnesite	9.359	AA_Magnesite -> 1 CO3-2 + 1 Mg+2	Fe2[MoO4]3[1]	82.02	Fe2[MoO4]3[1] + 8 H2O -> 2 Fe[OH]4+ + 8 H+ + 3 MoO4-2
AA_Portlandite	-22.80	AA_Portlandite + 2 H+ -> 1 Ca+2 + 2 H2O	FeMoO4[s]	45.00	FeMoO4[s] + 4 H2O -> 1 Fe[OH]4+ + 4 H+ + 1 MoO4-2 + 1 e-
AA_Tobermorite-I	23.86	AA_Tobermorite-I -> 2 Ca+2 + 0.8 H+ + 1.2 H2O + 2.4 H2SiO4-2	Ferrihydrite	16.71	Ferrihydrite + 1 H2O -> 1 Fe[OH]4+ + 1 H+
Albite[low]	85.27	Albite[low] + 8 H2O -> 1 Al[OH]4- + 6 H+ + 3 H2SiO4-2 + 1 Na+	Fluorite	10.96	Fluorite -> 1 Ca+2 + 2 F-
Antimocrandallite-exp	63.00	Antimocrandallite-exp + 8 H2O -> 3 Al[OH]4- + 1 Ca+2 + 3 H+ + 2 Sb[OH]6-	Friedel_salt_CN-HS	-15.77	Friedel_salt_CN-HS + 4 H+ -> 2 Al[OH]4- + 2 CN- + 4 Ca+2 + 4 H2O
Arsenocrandallite-therm	95.56	Arsenocrandallite-therm + 6 H2O -> 3 Al[OH]4- + 2 AsO4-3 + 1 Ca+2 + 7 H+	Manganite	-25.27	Manganite + 3 H+ + 1 e- -> 2 H2O + 1 Mn+2
Austinite-therm	11.47	Austinite-therm + 1 H+ -> 1 AsO4-3 + 1 Ca+2 + 1 H2O + 1 Zn+2	Molybdocrandallite-exp	86.00	Molybdocrandallite-exp + 6 H2O -> 3 Al[OH]4- + 1 Ca+2 + 7 H+ + 3 MoO4-2
Ba[Scr]O4[96%SO4]	9.790	Ba[Scr]O4[96%SO4] -> 1 Ba+2 + 0.04 CrO4-2 + 0.96 SO4-2	Ni[OH]2[s]	-10.80	Ni[OH]2[s] + 2 H+ -> 2 H2O + 1 Ni+2
beta-TCP	28.93	beta-TCP -> 3 Ca+2 + 2 PO4-3	Ni2SiO4	7.079	Ni2SiO4 + 2 H+ -> 1 H2SiO4-2 + 2 Ni+2
Boehmite	14.42	Boehmite + 2 H2O -> 1 Al[OH]4- + 1 H+	Pb[OH]2[C]	-8.150	Pb[OH]2[C] + 2 H+ -> 2 H2O + 1 Pb+2
Ca[OH]2.Sn[OH]2	-17.00	Ca[OH]2.Sn[OH]2 + 4 H+ -> 1 Ca+2 + 4 H2O + 1 Sn+2	PbMoO4[c]	15.80	PbMoO4[c] -> 1 MoO4-2 + 1 Pb+2
Ca2Cr[OH]7.3H2O	25.30	Ca2Cr[OH]7.3H2O + 1 H2O -> 2 Ca+2 + 1 CrO4-2 + 1 H+ + 3 e-	Sn[OH]2[s]	1.447	Sn[OH]2[s] + 2 H+ -> 2 H2O + 1 Sn+2
Ca3[AsO4]2:3H2O	21.40	Ca3[AsO4]2:3H2O -> 2 AsO4-3 + 3 Ca+2 + 3 H2O	ZnSiO3	18.69	ZnSiO3 + 1 H2O -> 1 H2SiO4-2 + 1 Zn+2
Ca3[OH]2[MoO4]2[c]	-3.000	Ca3[OH]2[MoO4]2[c] + 2 H+ -> 3 Ca+2 + 2 H2O + 2 MoO4-2			





Model Comparison: residuals - Concentration

Sample

Name Industrial CN site

Legend

Total Average Deviation Square root of the sum of the squared values of residuals divided by the number of values, over the entire X range.
User Average Deviation Square root of the sum of the squared values of residuals divided by the number of values, over the user defined X range.
Fractional Average Deviation Square root of the sum of the squared values of residuals divided by the number of values, over the fraction.
 Note that the Total and User Average Deviation columns are averages as well.

Residual details, concentrations

Residuals as log(model/sample)									
Fraction	8	7	6	5	4	3	2	1	Total Avg
pH									Deviation
Al	0.00	0.01	-1.87	-2.11	-2.38	0.10	-0.31	0.45	0.47
As	-0.12	1.27	-1.08	-1.46	-0.57	0.02	0.64	0.51	0.30
Ba	0.01	0.00	0.12	0.18	0.20	0.15	0.13	0.15	0.05
Br	0.27	0.28	0.05	0.04	0.00	0.02	0.10	0.10	0.05
Ca	0.03	0.00	0.01	0.07	0.10	-0.04	-0.32	-0.22	0.05
Cd	0.00	0.11	-0.31	0.48	0.32	0.13	0.54	0.30	0.11
Cl	-0.01	-0.04	-0.05	-0.04	-0.04	-0.02	0.00	0.00	0.01
CNT	0.12	0.27	0.02	0.29	0.16	1.06	0.83	-0.33	0.18
Co	0.00	-0.01	0.09	1.67	2.13	-0.52	0.57	1.11	0.38
CO32-	-	-	-	-	-	-	-	-	-
Cr	-1.54	-1.10	0.81	0.86	0.93	0.92	-0.20	0.53	0.33
Cu	-0.18	-0.51	2.09	1.97	0.59	1.73	0.80	1.05	0.46
F	0.00	0.05	-1.54	-0.89	-0.94	-0.31	-0.70	-0.69	0.28
Fe	-0.03	0.38	0.48	0.39	0.12	0.64	-0.01	0.38	0.13
B	0.00	0.09	0.38	0.55	1.78	2.08	1.13	-1.04	0.40
Si	0.00	0.36	0.67	0.84	1.22	1.35	0.62	1.29	0.32
K	0.01	-0.01	0.00	-0.04	0.02	0.00	-0.01	0.00	0.01
Li	0.01	0.00	0.07	0.12	0.08	0.27	0.01	-0.10	0.04
Mg	0.00	0.00	0.04	0.06	0.15	-1.22	-0.82	-1.27	0.24
Mn	0.00	0.00	0.14	0.56	2.58	3.94	0.91	0.55	0.61
Mo	0.69	1.89	0.17	0.04	0.00	0.55	0.11	-0.47	0.27
Na	0.03	-0.07	-0.07	-0.07	-0.07	-0.09	-0.01	-0.19	0.03
Ni	-0.01	0.05	0.26	0.54	0.65	-1.27	-0.11	1.73	0.29
Pb	-0.04	0.24	1.07	0.73	0.70	0.12	0.88	1.18	0.26
PO4	-	-	-	-	-	-	-	-	-
Sb	0.00	0.42	0.44	0.43	0.18	1.25	1.21	-0.48	0.25
Se	0.57	0.56	0.22	0.27	0.52	0.13	0.24	0.00	0.13
Sn	-0.01	0.72	0.21	0.22	0.26	1.07	-0.15	-1.45	0.25
SO4	0.09	0.08	0.44	0.51	0.54	0.94	0.39	-0.03	0.17
Sr	0.02	0.00	0.04	0.09	0.10	0.09	0.00	0.00	0.02
V	-1.47	-0.78	-2.24	-3.26	-0.91	-0.75	-0.59	-0.66	0.57
W	-1.10	0.63	0.67	0.67	0.67	1.18	0.70	1.10	0.31
Zn	0.00	0.13	-0.99	0.58	0.34	0.39	0.62	2.05	0.31
Avg Deviation	0.08	0.10	0.15	0.18	0.17	0.20	0.10	0.15	0.24

Yellow = own pH All residuals within + 1 or - 1 are considered to represent a good fit. 0 of course the best.

B Phase missing pH 8-12
 Sb Actually pretty good description for a difficult element
 CN Note description of CN with ettringite substitution!!!