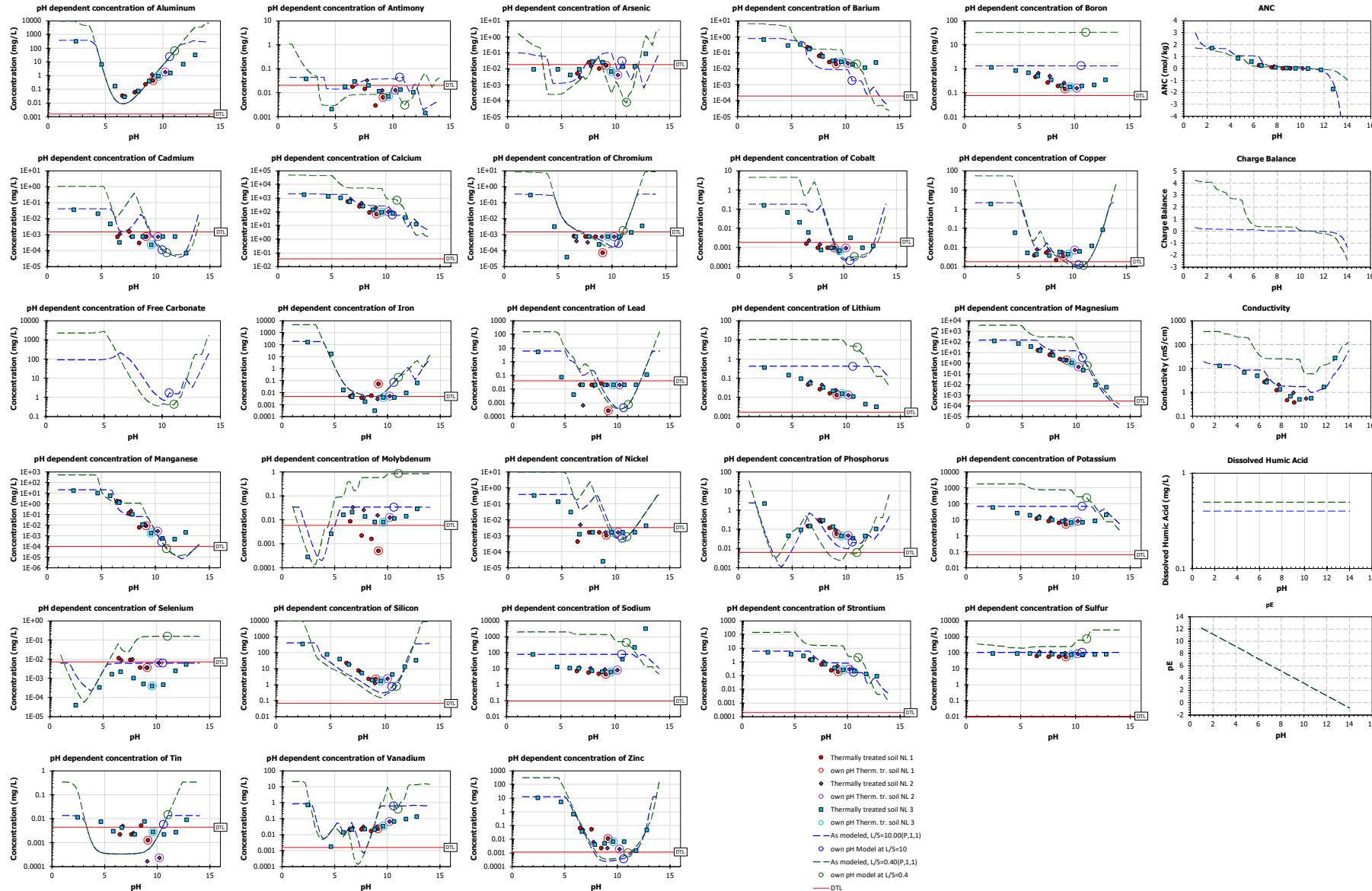


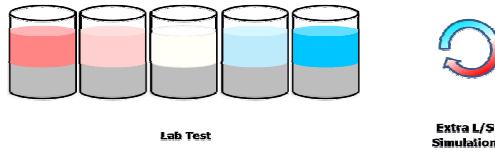
THERMALLY TREATED SOIL NL

COMPARISON pH DEPENDENCE AND MODEL



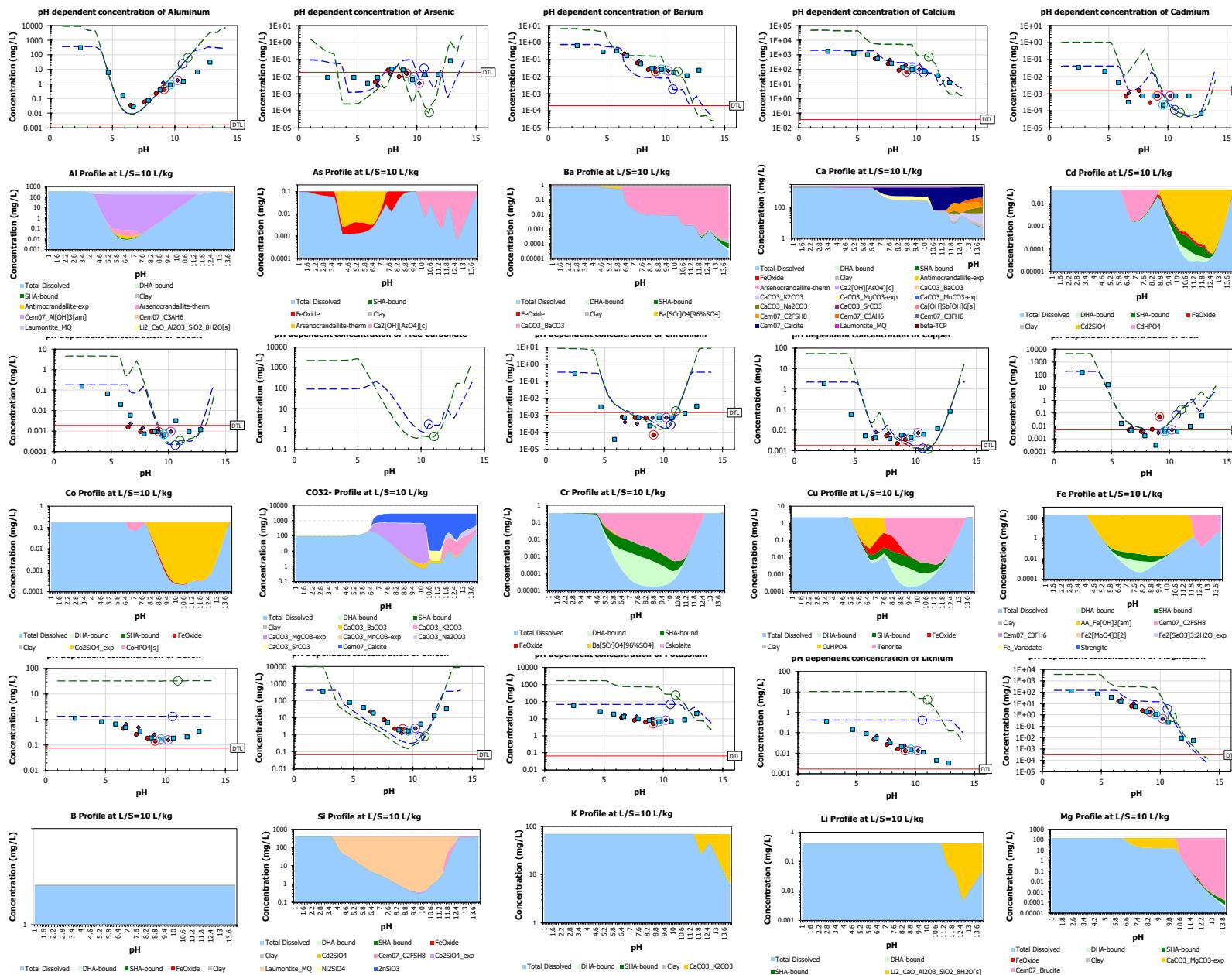
Object Name pH Dependent Leaching Test Model
Thermally treated soil NL

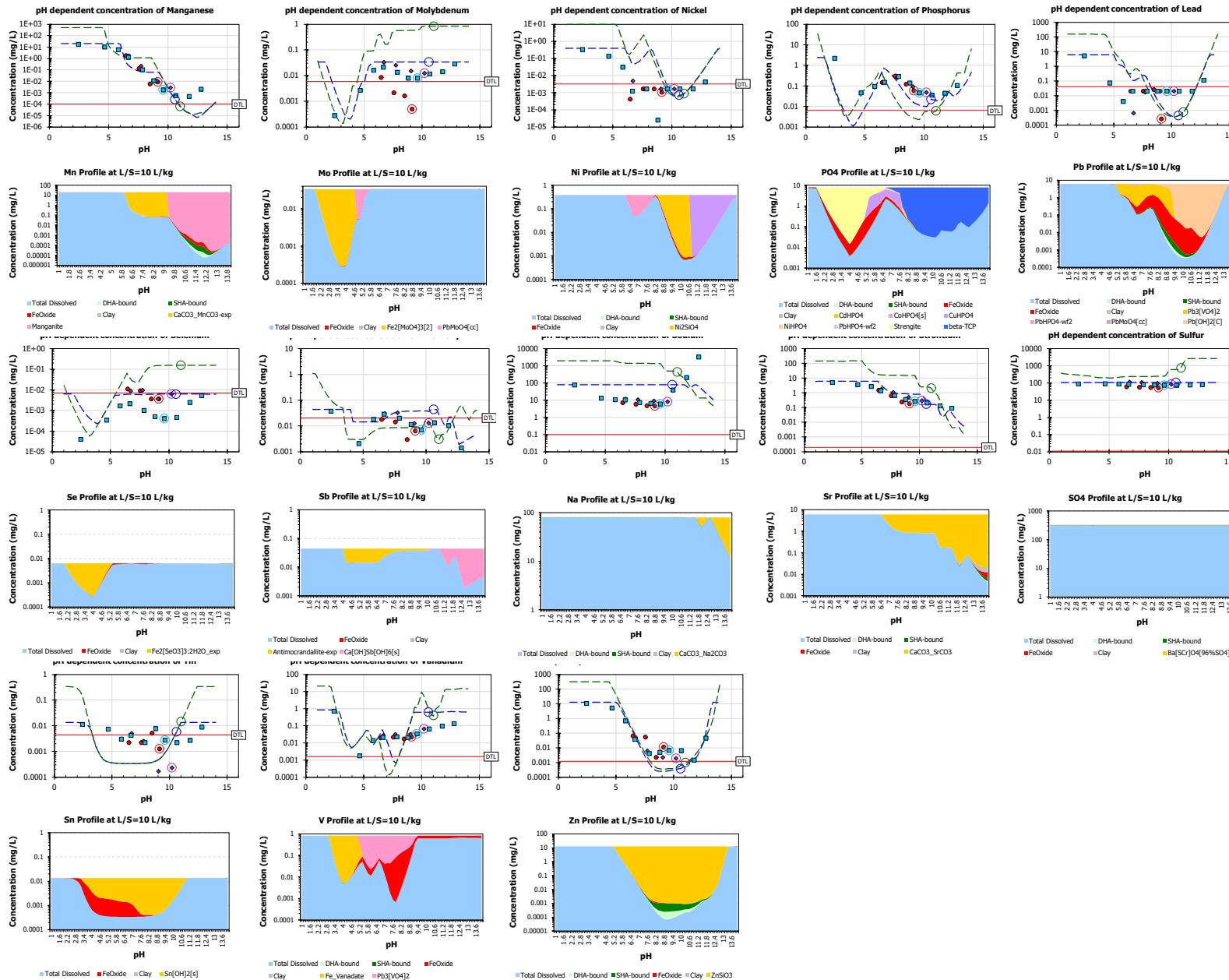
pH Dependent Leaching Test Scenario



Lab Test		Extra L/S Simulation								
Model Parameters	Entity	Unit	Default	Available Content	Entity	mg/kg	Entity	mg/kg	Entity	mg/kg
c0			-6.699	Acetic acid	F	2.220E-08	Pb	1.900E-09	Pb	61.18
c1			3.144E-11	Ag	Fe	1.079E-08	PO4	1859	PO4	79.50
c2			-9.796E-12	Al	B	3739	Sb	13.31	Sb	0.4414
c3			1.397E-12	As	Si	1.030	Se	4070	Se	0.06251
c4			-9.302E-14	Ba	Hg	7.838	Sn	2.006E-08	Sn	0.1347
c5			2.346E-15	Br	K	7.990E-09	SO4	696.2	SO4	3186
Clay	Clay	mg/kg	8000	Ca	Li	2.065E+04	Sr	4.203	Th	59.88
Hydrous Ferric Oxide		mg/kg	70.00	Cd	Mg	0.4203	U	1485	2.320E-08	
L/S		L/kg	10.00	Cl	Mn	100.0	V	204.0	2.380E-08	
pE			7.300	Co	Mo	1.817	Zn	0.3366		8.431
pH			5.860	CO32-	Na	2.785E+04		800.0		
Solid Humic Acid		mg/kg	20.00	Cr	Ni	3.440		3.861		
Simulated Low L/S		L/kg	0.4000	Cu	NO3	22.02		6.200E-09		125.0
Minerals										
Name	Name	Log(K)	Reaction		Name	Log(K)	Reaction			
AA_Fe[OH]3[am]	Cem07_CAH10	16.60	AA_Fe[OH]3[am] + 1 H2O -> 1 Fe[OH]4- + 1 H+		Cem07_Calcite	7.505	Cem07_CAH10 -> 2 Al[OH]4- + 1 Ca+2 + 6 H2O			
Antimocrandallite-exp	Cem07_Gypsum	63.00	Antimocrandallite-exp + 8 H2O -> 3 Al[OH]4- + 1 Ca+2 + 3 H+ + 2 Sb[OH]6-		Cerrusite	8.485	Cem07_Calcite -> 1 CO3-2 + 1 Ca+2			
Arsenocrandallite-ther	Ce02SiO4_exp	95.56	Arsenocrandallite-therm + 6 H2O -> 3 Al[OH]4- + 2 AsO4-3 + 1 Ca+2 + 7 H+		Co2SiO4_exp	4.583	Cem07_Gypsum -> 1 Ca+2 + 2 H2O + 1 SO4-2			
Ba[Scr]O4[96%SO4]	Cerrusite	9.790	Ba[Scr]O4[96%SO4] -> 1 Ba+2 + 0.04 CrO4-2 + 0.96 SO4-2		CoHPO4[s]	13.13	Cerrusite -> 1 CO3-2 + 1 Pb+2			
BaSrSO4[50%Ba]	CoHPO4	8.221	BaSrSO4[50%Ba] -> 0.5 Ba+2 + 1 SO4-2 + 0.5 Sr+2		CuHPO4	6.469	Co2SiO4_exp + 2 H+ -> 2 Co+2 + 1 H2SiO4-2			
beta-TCP	CuHPO4	28.93	beta-TCP -> 3 Ca+2 + 2 PO4-3		Eskolate	24.48	CoHPO4[s] -> 1 Co+2 + 1 H+ + 1 PO4-3			
Ca[OH]Sb[OH]6[s]	Eskolate	2.000	Ca[OH]Sb[OH]6[s] + 1 H+ -> 1 Ca+2 + 1 H2O + 1 Sb[OH]6-		Fe_Vanadate	26.00	CuHPO4 -> 1 Cu+2 + 1 H+ + 1 PO4-3			
Ca2[OH][AsO4][c]	Fe2[MoO4]3[2]	4.000	Ca2[OH][AsO4][c] + 1 H+ -> 1 AsO4-3 + 2 Ca+2 + 1 H2O		Fe2[SeO3]3:2H2O_exp	139.5	Eskolate + 5 H2O -> 2 CrO4-2 + 10 H+ + 6 e-			
Ca5[OH][VO4]3[c]	Fe2[SeO3]3:2H2O	-61.00	Ca5[OH][VO4]3[c] + 13 H+ -> 5 Ca+2 + 7 H2O + 3 VO2+		Laumontite_MQ	19.18	Fe_Vanadate + 1 H2O -> 0.5 Fe[OH]4- + 1 VO2+ + 0.5 e-			
CaCO3_BaCO3	Laumontite_MQ	22.00	CaCO3_BaCO3 -> 1 Ba+2 + 2 CO3-2 + 1 Ca+2		Fe2[MoO4]3[2]	-86.35	Fe2[MoO4]3[2] + 8 H2O -> 2 Fe[OH]4- + 8 H+ + 3 MoO4-2			
CaCO3_K2CO3	Manganite	19.30	CaCO3_K2CO3 -> 2 CO3-2 + 1 Ca+2 + 2 K+		Fe2[SeO3]3:2H2O	180.0	Fe2[SeO3]3:2H2O_exp + 7 H2O -> 2 Fe[OH]4- + 14 H+ + 3 SeO4-2 + 6 e-			
CaCO3_Li2CO3	Manganite	21.30	CaCO3_Li2CO3 -> 2 CO3-2 + 1 Ca+2 + 2 Li+		Manganite	-25.27	Laumontite_MQ + 8 H2O -> 2 Al[OH]4- + 1 Ca+2 + 8 H+ + 4 H2SiO4-2			
CaCO3_MgCO3-exp	Ni[OH]2[s]	18.02	CaCO3_MgCO3-exp -> 2 CO3-2 + 1 Ca+2 + 1 Mg+2		Ni[OH]2[s]	-10.80	Manganite + 3 H+ + 1 e- -> 2 H2O + 1 Mn+2			
CaCO3_MnCO3-exp	Ni2SiO4	20.78	CaCO3_MnCO3-exp -> 2 CO3-2 + 1 Ca+2 + 1 Mn+2		Ni2SiO4	5.498	Ni[OH]2[s] + 2 H+ -> 2 H2O + 1 Ni+2			
CaCO3_Na2CO3	NiHPO4	18.30	CaCO3_Na2CO3 -> 2 CO3-2 + 1 Ca+2 + 2 Na+		NiHPO4	25.00	Ni2SiO4 + 2 H+ -> 1 H2SiO4-2 + 2 Ni+2			
CaCO3_SrCO3	Pb[OH]2[C]	19.85	CaCO3_SrCO3 -> 2 CO3-2 + 1 Ca+2 + 1 Sr+2		Pb[OH]2[C]	-25.27	NiHPO4 -> 1 H+ + 1 Ni+2 + 1 PO4-3			
CaMoO4[s]	Pb3[VO4]2	7.950	CaMoO4[s] -> 1 Ca+2 + 1 MoO4-2		Pb3[VO4]2	-10.80	Pb[OH]2[C] + 2 H+ -> 2 H2O + 1 Pb+2			
CaSb[OH]6[s]2_exp	PbHPO4-wf2	19.41	CaSb[OH]6[s]2_exp -> 1 Ca+2 + 2 Sb[OH]6-		PbHPO4-wf2	5.498	Pb3[VO4]2 + 8 H+ -> 4 H2O + 3 Pb+2 + 2 VO2+			
Cd2SiO4	PbMoO4[cc]	6.059	Cd2SiO4 + 2 H+ -> 2 Cd+2 + 1 H2SiO4-2		PbMoO4[cc]	25.48	PbHPO4-wf2 -> 1 H+ + 1 PO4-3 + 1 Pb+2			
CdHPO4	Strengite	26.48	CdHPO4 -> 1 Cd+2 + 1 H+ + 1 PO4-3		Sn[OH]2[s]	-25.27	PbMoO4[cc] -> 1 MoO4-2 + 1 Pb+2			
Cem07_Al[OH]3[am]	Tenorite	13.76	Cem07_Al[OH]3[am] + 1 H2O -> 1 Al[OH]4- + 1 H+		Strengite	-8.150	Sn[OH]2[s] + 2 H+ -> 2 H2O + 1 Sn+2			
Cem07_Brucite	ZnSiO3	-16.83	Cem07_Brucite + 2 H+ -> 2 H2O + 1 Mg+2		Tenorite	48.00	Strengite + 2 H2O -> 1 Fe[OH]4- + 4 H+ + 1 PO4-3			
Cem07_C2FSH8	ZnSiO3	21.41	Cem07_C2FSH8 -> 2 Ca+2 + 2 Fe[OH]4- + 3 H2O + 1 H2SiO4-2		ZnSiO3	-7.620	Tenorite + 2 H+ -> 1 Cu+2 + 1 H2O			
Cem07_C3AH6		-35.14	Cem07_C3AH6 + 4 H+ -> 2 Al[OH]4- + 3 Ca+2 + 4 H2O			18.69	ZnSiO3 + 1 H2O -> 1 H2SiO4-2 + 1 Zn+2			
Cem07_C3FH6		-30.82	Cem07_C3FH6 + 4 H+ -> 3 Ca+2 + 2 Fe[OH]4- + 4 H2O							

COMPOSITION AND PARTITIONING





Model Comparison: residuals - Concentration

Name Thermally treated soil NL

Legend

- Total Average Deviation** Square root of the sum of the squared values of residuals divided by the number of values, over the entire X range.
 - User Average Deviation** Square root of the sum of the squared values of residuals divided by the number of values, over the user defined X range.
 - Fractional Average Deviation** Square root of the sum of the squared values of residuals divided by the number of values, over the fraction.
- Note that the Total and User Average Deviation columns are averages as well.

Residual details, concentrations

Residuals as log(model/sample)

Fraction	10	9	8	7	6	5	4	3	2	1	Total Avg	
pH	2.44	4.66	5.82	6.65	7.86	8.80	9.60	10.7	11.8	12.8	0.25	Yellow = own pH All residuals within + 1 or - 1 are considered to represent a good fit.
Al	0.07	-0.02	-0.96	-0.48	-0.18	-0.01	0.43	1.24	1.47	1.03	0.25	As Measurements at MDL
As	0.90	-0.87	-0.43	-0.35	-0.29	0.54	1.18	0.28	-0.05	-1.84	0.27	K Both Na and K show similar pH dependence, which may be linked to dissolution of silica phases. Otherwise no indication what is causing this behaviour
B	0.07	0.21	0.29	0.42	0.59	0.83	0.88	0.85	0.80	0.59	0.20	Li Strong pH dependent behaviour. As yet no phases identified that can explain the behaviour over the wide pH range where solubility control is active.
Ba	0.07	0.36	0.20	-0.63	-0.77	-0.53	-0.50	-1.01	-1.44	-1.76	0.28	Possibly linked to Si phases?
Ca	0.07	0.16	0.27	0.35	0.08	0.32	0.46	-0.11	-0.43	0.21	0.09	Mg Further look into carbonate phases
Cd	0.07	0.30	0.94	0.76	0.72	1.30	0.58	-0.85	-1.21	0.13	0.25	Na Both Na and K show similar pH dependence, which may be linked to dissolution of silica phases. Otherwise no indication what is causing this behaviour
Cl	-	-	-	-	-	-	-	-	-	-	-	Pb Measurements at MDL
Co	0.07	0.43	0.95	1.16	2.26	0.65	-0.29	-1.18	-0.48	0.06	0.31	Se Difficult element to capture due to lacking thermodynamic data. Experimental minerals may need to be adjusted.
CO32-	-	-	-	-	-	-	-	-	-	-	-	V Reasonable description. Info lacking at pH > 9
Cr	0.06	1.83	2.11	0.46	0.10	0.32	-0.47	-0.36	1.35	1.99	0.38	Zn Good description. Measurements at own pH at MDL
Cu	0.07	1.57	1.36	0.21	0.46	-0.23	-0.35	-0.69	-0.49	0.09	0.23	
F	-	-	-	-	-	-	-	-	-	-	-	
Fe	0.06	-0.44	0.58	0.48	0.55	1.20	0.38	1.33	2.06	0.54	0.30	
K	0.07	0.42	0.57	0.72	0.85	0.94	1.02	1.00	0.64	0.23	0.23	
Li	0.07	0.46	0.66	0.89	1.11	1.29	1.43	1.57	1.97	2.10	0.41	
Mg	0.07	0.32	0.60	0.64	0.50	0.89	1.13	1.05	0.35	-0.89	0.23	
Mn	0.07	0.31	0.54	-0.68	-0.17	0.74	0.89	-0.41	-1.43	-2.30	0.31	
Mo	1.04	0.18	0.31	0.20	0.39	0.62	0.62	0.47	0.38	0.07	0.16	
Na	0.01	0.79	0.87	0.88	1.04	1.10	1.11	0.30	-0.51	-1.70	0.30	
Ni	0.07	0.46	1.10	1.59	2.10	3.32	0.19	-0.36	0.30	0.85	0.45	
Pb	0.07	1.92	2.38	0.72	0.89	-0.63	-1.47	-1.65	-0.40	0.76	0.41	
PO4	-	-	-	-	-	-	-	-	-	-	-	
Sb	0.07	0.85	-0.10	-0.21	0.25	0.51	0.74	0.51	0.30	0.19	0.14	
Se	1.56	0.53	0.54	0.45	0.76	1.10	1.19	1.13	0.41	0.07	0.28	
Si	0.07	-0.34	-0.63	-0.66	-0.57	-0.55	-0.72	-0.73	0.07	1.03	0.19	
Sn	0.05	-1.28	-0.95	-1.10	-0.80	-1.26	-0.50	0.47	0.69	0.18	0.26	
SO4	-	-	-	-	-	-	-	-	-	-	-	
Sr	0.07	0.22	0.34	0.42	0.15	0.38	0.49	-0.09	-0.52	-0.34	0.11	
V	0.07	0.96	0.02	0.38	-1.43	0.37	1.26	0.97	0.82	0.70	0.26	
Zn	0.07	0.37	0.22	0.22	-0.66	-1.31	-1.38	-1.22	0.08	0.44	0.24	
Avg Deviation	0.08	0.16	0.18	0.14	0.18	0.21	0.18	0.18	0.19	0.21	0.26	